



# Chemifloc Ltd.

## SAFETY DATA SHEET Sodium Carbonate

Conforms to Regulation (EC) No. 1907/2006 (REACH), Annex II, Regulation (EC) 1272/2008 and Regulation (EC) 453/2010.

### Section 1: Identification of the substance and of the company/undertaking

#### Identification of the substance or mixture

**Product Name:** Sodium Carbonate  
**Chemical Name:** Sodium Carbonate  
**Registration Number:** 01-2119485498-19-0018  
**Synonyms:** Disodium carbonate, soda ash  
**Chemical Formula :** Na<sub>2</sub>CO<sub>3</sub>  
**Date of first issue:** 26-09-2011  
**Version number** 03  
**Revision date:** 24-03-2016  
**Supersedes date:** 04-03-2016

#### Relevant identified uses of the substance or mixture and uses advised against:

**Identified uses**  
- Treatment for potable water, industrial water and residual water.  
- Washing and cleaning products.

**Uses advised against** None

#### Details of the supplier of the safety data sheet

**Supplier:** Chemifloc Ltd  
Smithstown, Shannon,  
Co. Clare,  
Rep. of Ireland.  
Tel: 00353 61 708699  
Fax: 00353 61 708698  
e-mail: [info@chemifloc.ie](mailto:info@chemifloc.ie)

**Emergency Telephone Number: National Poison Information Centre,  
00353 1 8379964**

### Section 2: Hazards Identification

#### Classification of the substance

The substance has been assessed and/or tested for its physical, health and environmental hazards and the following classificatory applies.

#### Classification according to Regulation (EC) no 1272/2008 as amended

Eye Irritant 2

#### Label elements

#### Label according to Regulation (EC) No. 1272/2008 as amended

**Contains:** Sodium Carbonate



**Signal word**

Warning

**Hazard statements**

H319: Causes serious eye irritation.

### Precautionary statements

P264: Wash hands and face thoroughly after handling .  
P280: Wear protective gloves, protective clothing, eye protection and face protection.  
P305+P351+P338: IF IN EYES: Rinse cautiously with water for several minutes.  
Remove contact lenses, if present and easy to do. Continue rinsing.  
P337+P313: If eye irritation persists: Get medical advice/attention.

### Other hazards

The substance does not meet the criteria for PBT or vPvB (see section 12).  
No other hazards identified.

## Section 3: Composition/Information on Ingredients

### Substance:

Name: Sodium Carbonate

Name	EC number	CAS number	Concentration	Classification Directive	Classification Regulation (EC) 1272/2008
Sodium carbonate	207-838-8	497-19-8	>99%	H319	Eye Irritant 2

## Section 4: First Aid Measures

### Description of first aid measures

#### General information:

No known delayed effects

#### In case of inhalation:

Remove to fresh air, keep warm and at rest.  
If symptoms persist, seek medical attention.

#### After skin contact:

Remove contaminated clothing and wash before re-use.  
Wash off with soap and water.  
If symptoms persist, seek medical attention.

#### After eye contact:

Remove contact lenses if present.  
Irrigate eye thoroughly with eye wash solution or clean water for at least 15 minutes.  
Eyelids should be held away from the eyeball to ensure thorough rinsing.  
If eye irritation persists seek medical attention.

#### In case of ingestion:

DO NOT induce vomiting.  
Wash out mouth with water and give plenty of water to drink (at least 300 ml.).  
Obtain medical advice if necessary.

## Section 5: Firefighting measures

### Extinguishing media:

#### Suitable extinguishing media:

Use extinguishing measures that are appropriate to local circumstances and the surrounding environment.

#### Unsuitable extinguishing media:

None.

#### Special hazards arising from the substance or mixture

None.

#### Advice for firefighters

No special precautions required.

## Section 6: Accidental release measures

### Personal Precautions:

#### For non-emergency personnel:

Keep dust levels to a minimum.

Wear suitable protective equipment ( see Section 8).

### Environmental Precautions:

Prevent uncontrolled discharges into the environment (rivers, water courses, sewers etc.).

Avoid any mixture with an acid into sewer/drains (CO<sub>2</sub> gas formation).

### Methods for containment and clean up:

In all cases avoid dust formation.

Use vacuum suction, or shovel into bags.

Collect as much as possible in a suitable clean container, preferably for re-use, otherwise for disposal (See Section 13).

### Reference to other sections:

For more information on exposure controls/personal protection or disposal considerations, please see section 8 and 13.

## Section 7: Handling and storage

### Precautions for Safe Handling:

#### Protective measures:

Keep dust levels to a minimum.

Ensure adequate ventilation.

Wear protective equipment (see Section 8.2).

Keep away incompatible materials.

#### Advice on general occupational hygiene:

Good personal and housekeeping practices to be used.

No drinking, eating or smoking at the workplace.

#### Conditions for safe storage, including any incompatibilities:

Store in a dry place.

Store in original, closed and correctly labelled container.

Store away from incompatible materials.

## Section 8: Exposure controls / personal protection

### Control Parameters:

#### Occupational Exposure Standards:

Not listed by H&SE (Guidance Note EH40) or ACGIH.

#### Recommended Limits:

WEL 10mg/m<sup>3</sup> (total dust) (8hr TWA).

4mg/m<sup>3</sup> (respirable dust) (8hr TWA).

#### DNEL's/PNEC:

Exposure route of relevance	DNELs (local effects)			
	Workers		General population	
	Long term	Acute	Long term	Acute
Inhalation	10 mg/m <sup>3</sup>			

#### PNEC:

The lowest L(E)C<sub>50</sub> value is > 100 mg/l (48h EC<sub>50</sub> is 200 mg/l in daphnids (Ceriodaphnia sp)). Therefore sodium carbonate need not be classified according to Directive 67/548/EEC and EU Classification, Labelling and Packaging of Substances and Mixtures (CLP) Regulation (EC) No. 1272/2008.

**Environmental Classification is not warranted.**

#### Exposure Controls:

Appropriate engineering controls É provide appropriate exhaust ventilation at places where dust is formed.

Apply technical measures to comply with the occupational exposure limits.

**Personal protection:**

**Eye/face protection:**

Wear eye/face protection rated to protect eyes against dust (EN166) eg. safety eye shields with dust protection, goggles or face visor

**Hand protection:**

Wear suitable chemical resistant protective gloves that comply with the specification of EC Directive 89/686/EEC and the related standard EN374. Suitable materials, Neoprene or natural rubber.

**Skin/body protection:**

Dust impervious protective suit, rubber or plastic safety boots.

**Respiratory protection:**

In the case of high dust levels wear suitable respiratory protective equipment eg. dust mask or respirator, that conform to national/international standard, EN143. Recommended filter type P2.

**Environmental Exposure Controls:**

Contain any spillage.

Avoid discharges to the environment.

Dispose of any rinse water in accordance with local and national regulations.

## Section 9: Physical and chemical properties

### Information on basic physical and chemical properties

#### General information (Appearance, odour)

Physical State	White Powder
Odour	Odourless

#### Important health safety and environmental information

pH	>11 (saturated solution, study result, OECD Guideline105).
Melting/freezing point	851°C
Boiling point	not applicable (melting point >300°C)
Flash point	Non flammable
Flammability (solid, gas)	not applicable
Explosive properties	
- Lower explosive limit	Non explosive.
- Upper explosive limit	
Oxidising Properties	Non oxidizing.
Vapour Pressure	Not applicable.
Relative Density	2.52 @ 20°C (study result, EU Method A.3)
Viscosity	Not applicable.
Solubility(ies)	
Water solubility	212.5 g/l @20°C. (study result, OECD Guideline 105)

## Section 10: Stability and reactivity

### Reactivity

Decomposes by reaction with strong acids to evolve carbon dioxide.

### Chemical stability

Stable under recommended storage conditions (see Section 7).

### Possibility of hazardous reactions

None.

### Conditions to avoid

Contact with acids unless under controlled conditions.

Exposure to moisture.

### Incompatible materials

Finely divided aluminium.

### Hazardous decomposition products

None.

## Section 11: Toxicological information

**Information on Toxicological Effects:**

Toxicity endpoints	Details of the effects assessment
<b>Acute toxicity</b>	<p>Oral : LD50, rat 2800mg/kg bw                      Dermal :LD50, rabbit &gt;2000mg/kg bw, Method: EPA 16 CFR 1500.40                      Inhalation : LC50, rat 2300 mg/m<sup>3</sup>air Method: based on OECD Guideline 403</p> <p>Values exceed the cut off limit of 2000mg/kg established by EU Directive 67/548/EEC and EU Classification, Labelling and Packaging of Substances and Mixtures (CLP) Regulation (EC) No. 1272/2008                      Classification for acute toxicity: is not warranted</p>
<b>Irritation/Corrosion</b>	<p>Eye Irritation : irritating Method: OECD Guideline 405                      Skin Irritation : not irritating Method: OECD Guideline 404                      Respiratory irritation : not irritating Based on available data</p> <p>Classification for Eye irritancy : Xi, R36 (irritating to eyes) according to Directive 67/548/EEC.                      :Category 2, H319 (causes serious eye irritation) according to CLP Regulation (EC) 1272/2008.                      Classification for Skin irritancy : is not warranted.                      Classification for Respiratory irritancy : is not warranted.</p>
<b>Sensitisation</b>	<p>No data available on the sensitisation of sodium carbonate. Sodium carbonate is considered not to have any sensitising properties, based on the physiological role of both its constituent ions and its longterm historical and wide dispersive use in industrial processes and consumer products.                      Classification for sensitisation : is not warranted.</p>
<b>Repeated dose toxicity</b>	<p>Oral : Sodium carbonate dissociates into ions that are present physiologically in relatively high levels in vertebrates. Therefore, repeated dose toxicity studies are considered (scientifically) unnecessary, in accordance with column 2 of REACH Annex VIII and IX. Furthermore, sodium carbonate is used as a food additive, which confirms that the substance has a low Repeated dose toxicity.</p> <p>Dermal : Sodium carbonate dissociates into ions that are present physiologically in relatively high levels in vertebrates. Therefore, repeated dose toxicity studies are considered (scientifically) unnecessary, in accordance with Column 2 of REACH Annex VIII and IX</p> <p>Inhalation : Sodium carbonate dissociates into ions that are present physiologically in relatively high levels in vertebrates. Therefore, repeated dose toxicity studies are considered (scientifically) not necessary, In accordance with column of REACH Annex VIII and IX.</p> <p>Classification for repeated dose toxicity: is not warranted</p>
<b>Mutagenicity</b>	<p>In vitro <math>\sigma</math>The available in vitro tests (SOS chromotest with sodium carbonate and Ames test with sodium bicarbonate) were negative. Furthermore sodium bicarbonate is naturally present in cells and both the structure of sodium bicarbonate and sodium carbonate do not indicate a genotoxic potential. Therefore, there is no reason to evaluate the potential genotoxicity of sodium carbonate further and no genotoxic effects are expected.                      Classification for mutagenicity is not warranted.</p>
<b>Carcinogenicty</b>	<p>No data available for carcinogenicity of sodium carbonate. Although the substance has a wide and varied use, there are no indications that it can induce hyperplasia, pre-neoplastic lesions or is mutagenic. Therefore, a carcinogenicity study is considered unnecessary.                      Classification for carcinogenicity is not warranted</p>
<b>Reproductive toxicity</b>	<p>Fertility : No data available                      Developmental toxicity : In accordance with Section 1 of REACH Annex XI, testing does not appear scientifically necessary, as the substance will usually not reach the foetus or the male and female reproductive organs when exposed orally, dermally or by inhalation, as it does not become available systemically. As such, it is considered not useful to perform a reproduction study.                      Classification for reproductive toxicity according to Regulation (EC) 1272/2008 is not required</p>

## **Toxicity**

### **Acute toxicity to fish**

LC<sub>50</sub> (96hr) for freshwater fish : 300 mg/l

### **Chronic toxicity to fish**

Study scientifically unjustified, sodium carbonate dissociates readily into sodium and carbonate ions in an aquatic environment. Both ions originally exist in nature, and their concentrations in surface water are dependent on various factors, such as geological parameters, weathering and human activities.

Therefore, there is a continuous source of both ions into the environment and have been measured extensively in aquatic ecosystems.

### **Acute toxicity to crustaceans**

EC50 (48h) for freshwater invertebrates : 200-227 mg/l

### **Chronic toxicity to crustaceans**

Study scientifically unjustified, sodium carbonate dissociates readily into sodium and carbonate ions in an aquatic environment. Both ions originally exist in nature, and their concentrations in surface water are dependent on various factors, such as geological parameters, weathering and human activities. Therefore, there is a continuous source of both ions into the environment and have been measured extensively in aquatic ecosystems.

### **Acute toxicity to algae and other aquatic plants**

Study scientifically unjustified, sodium carbonate dissociates readily into sodium and carbonate ions in an aquatic environment. Both ions originally exist in nature, and their concentrations in surface water are dependent on various factors, such as geological parameters, weathering and human activities. Therefore, there is a continuous source of both ions into the environment and have been measured extensively in aquatic ecosystems.

### **Toxicity to soil macro-organisms**

In accordance with REACH Annex XI a study is not required as in water sodium carbonate is dissociated into sodium and carbonate ions, both of which will not adsorb on particulate matter. Furthermore, exposure of the soil compartment is unlikely.

### **Toxicity to terrestrial plants**

In accordance with REACH Annex XI a study is not required as in water sodium carbonate is dissociated into sodium and carbonate ions, both of which will not adsorb on particulate matter. Furthermore, exposure of the soil compartment is unlikely.

### **Persistence and degradability**

In water : Not applicable (quickly dissociates)

In soil : Not applicable (inorganic substance)

In sediment : Not applicable (inorganic substance)

### **Bioaccumulative Potential**

Not bioaccumulative (inorganic substance that in water dissociates into sodium and carbonate ions, which do not accumulate in living tissues).

### **Mobility in Soil**

If sodium carbonate is emitted to soil it can escape to atmosphere as carbon dioxide, precipitate as a metal carbonate, form complexes or stay in solution.

### **Results of PBT and vPvB Assessment**

According to Annex XIII of REACH Regulation inorganic substances do not require assessment.

### **Other Adverse Effects**

No other adverse effects are identified.

## Section 13: Disposal considerations

### Waste treatment methods

- If recycling or re-use is not practicable, dispose of in compliance with local or national regulations.
- Neutralise with acid under controlled conditions.
- Dilute with plenty of water

#### Packaging:

- Where possible, recycling is preferred to disposal or incineration.
- Clean container with water, dispose of rinse water in accordance with local or national regulations.
- Must be incinerated in a registered incineration plant with permit from the local authorities.

## Section 14: Transport information

**Sodium carbonate is not classified as hazardous for transport.**

**ADR** Not classified as dangerous in the meaning of transport regulations.

**IATA** Not classified as dangerous in the meaning of transport regulations.

**IMDG** Not classified as dangerous in the meaning of transport regulations.

## Section 15: Regulatory information

### Safety, health and environmental regulations/legislation specific for the substance or mixture

#### EU Regulations

Council Directive 96/82/EC on the control of major-accident hazards involving dangerous substances.

Council Directive 98/24/EC of 7 April 1998 on the protection of the health and safety of workers from the risks related to chemical agents at work.

#### Chemical Safety Assessment

A Chemical Safety Assessment has been carried out for the components of this substance.

#### Inventory status

Country(s) or region	Inventory name	On inventory (yes/no)*
Australia	Australian Inventory of Chemical Substances (AICS)	Yes
Canada	Domestic Substances List (DSL)	Yes
Canada	Non-Domestic Substances List (NDSL)	No
China	Inventory of Existing Chemical Substances in China (IECSC)	Yes
Europe	European Inventory of New and Existing Chemicals (EINECS)	Yes
Europe	European List of Notified Chemical Substances (ELINCS)	No
Japan	Inventory of Existing and New Chemical Substances (ENCS)	Yes
Korea	Existing Chemicals List (ECL)	Yes
New Zealand	New Zealand Inventory	Yes
Philippines	Philippine Inventory of Chemicals and Chemical Substances(PICCS)	Yes
United States & Puerto Rico	Toxic Substances Control Act (TSCA) Inventory	Yes

A "Yes" indicates that all components of this product comply with the inventory requirements administered by the governing country(s)

## Section 16: Other information

### Full text of H-Statements referred to under sections 2 and 3.

H319 Causes serious eye irritation.

**Training advice** Not available

### Further information

The information provided in this Safety Data Sheet is correct to the best of our knowledge, information and belief at the date of its publication. The information given is designed only as a guidance for safe handling, use, processing, storage, transportation, disposal and release and is not to be considered a warranty or quality specification. The information relates only to the specific material designated and may not be valid for such material used in combination with any other materials or in any process, unless specified in the text.

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